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Neutron diffraction study of multiferroic $\text{Tb}_{0.85}\text{Na}_{0.15}\text{MnO}_{3-y}$

T.S. Chan^a, R.S. Liu^{a,*}, C.C. Yang^{b,c}, W.-H. Li^c, Y.H. Lien^d, C.Y. Huang^e, J.W. Lynn^f

^aDepartment of Chemistry and Center for Nano Storage Research, National Taiwan University, Taipei 106, Taiwan

^bDepartment of Physics, National Central University, Chung-Li 320, Taiwan

^cInstitute of Physics, Academia Sinica, Taipei 115, Taiwan

^dInstitute of Materials Manufacturing, Chinese Culture University, Taipei 111, Taiwan

^eNational Taiwan Normal University, Taipei 106, Taiwan

^fNIST Center for Neutron Research, National Institute of Standards and Technology, Gaithersburg, MD 20899, USA

Abstract

The magnetic properties and thermal evolution of $\text{Tb}_{0.85}\text{Na}_{0.15}\text{MnO}_{3-y}$ have been studied by neutron-diffraction experiments. By comparison with the TbMnO_3 sample, the results show not only small lattice parameters but also a reduction in correlation length (ξ). This is strongly correlated to the induced oxygen vacancies in $\text{Tb}_{0.85}\text{Na}_{0.15}\text{MnO}_{3-y}$ [$y \approx 0.15(2)$].

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1. Introduction

The spin arrangement and magnetic properties in rare-earth manganites RMnO_3 were studied a long time ago. The ordering temperature for the Mn ions decreases with R ionic radius, ranging from $T_N = 141$ K for La to 40 K for Tb. Moreover, the R^{3+} cations also become magnetically ordered, but at low temperature [1–5]. On the other hand, the recent discovery of very large magnetoelectric effects (ME) in the TbMnO_3 has reopened the field of the so-called multiferroic materials [6,7]. Since the multiferroic materials with the coexistence of (anti)ferromagnetic and (anti)ferroelectric properties are one of the best candidates to enhance the ME effects. For TbMnO_3 , the incommensurate magnetic structure of the Mn^{3+} moments is well-known [8–13], which may be characterized by a modulation vector $(0, q_m, 0)$ that propagates along the second longest crystallographic direction b . q_m is incommensurate at the ordering temperature T_N (~ 42 K). In order to compare the differences between the magnetic properties and spin arrangement for TbMnO_3 , this paper is to investigate

the $\text{Tb}_{0.85}\text{Na}_{0.15}\text{MnO}_{3-y}$ magnetic structure and analyzes thermal evolution by using neutron powder diffraction (NPD).

2. Experimental procedures

The polycrystalline $\text{Tb}_{0.85}\text{Na}_{0.15}\text{MnO}_{3-y}$ sample was prepared by solid state reaction. Stoichiometric mixtures of Tb_4O_7 , Na_2CO_3 and MnCO_3 were sintered in air at 1450°C for 24 h. The high-resolution NPD patterns were collected on the BT-1 powder diffractometer at the NIST Center for Neutron Research, using a Cu (311) monochromator crystal and $15'-20'-7'$ FWHM angular collimations. These diffraction patterns were analyzed using the GSAS program [14].

3. Results and discussion

The $\text{Tb}_{0.85}\text{Na}_{0.15}\text{MnO}_{3-y}$ crystal structure was refined from a NPD pattern obtained at 300 K, with $\lambda = 1.5401 \text{ \AA}$ as shown in Fig. 1a. All the reflections were indexed in the orthorhombic space group (Pbnm) with the lattice parameters of $a = 5.2974(2) \text{ \AA}$, $b = 5.7775(3) \text{ \AA}$ and $c = 7.4151(3) \text{ \AA}$. These values are small with a recent

*Corresponding author. Tel.: +886 2 33661169; fax: +886 2 23693121.
E-mail address: rslu@ntu.edu.tw (R.S. Liu).

report of a bulk TbMnO_3 sample by Blasco et al. [15]. It is well known that the ionic radius of Na^+ is $\sim 10\%$ larger than that of Tb^{3+} . To be noted here is that the refined of oxygen occupancy obtained for the $\text{Tb}_{0.85}\text{Na}_{0.15}\text{MnO}_{3-y}$ at 300 K is essentially oxygen stoichiometric, i.e. $y \approx 0.15(2)$. Furthermore, bond valence sum calculations [16], based on the refined bond lengths, show that the valence of Mn ions in $\text{Tb}_{0.85}\text{Na}_{0.15}\text{MnO}_{3-y}$ is trivalent. These results indicate that the decrease in the oxygen content completely counterbalanced the effect of monovalent Na^+ for trivalent Tb^{3+} cation substitution. Therefore, the reduction of the lattice parameters in our sample is thus mainly due to the occurrence of oxygen deficiencies.

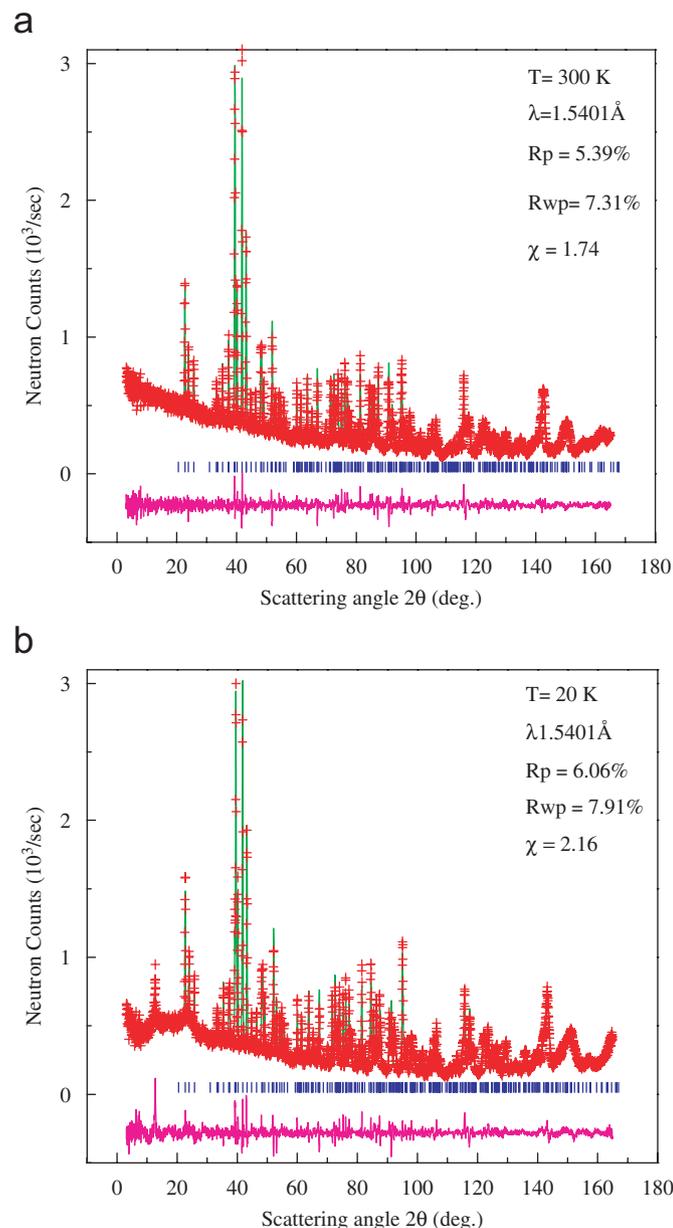


Fig. 1. Observed (crosses), calculated (solid line), and differences (bottom) NPD Rietveld profiles of $\text{Tb}_{0.85}\text{Na}_{0.15}\text{MnO}_{3-y}$ at (a) $T = 300 \text{ K}$ and (b) $T = 20 \text{ K}$. Bragg reflections are indicated by tick mark.

Fig. 1b shows experimental, calculated and different NPD patterns for $\text{Tb}_{0.85}\text{Na}_{0.15}\text{MnO}_{3-y}$ at 20 K. Rietveld analysis afforded sufficiently low R factors, suggesting that our structure analyses were successful. At $T = 20 \text{ K}$ new reflections are observed at 2θ angles not allowed for the Bragg positions in the space group Pbnm . This reveals the appearance of a magnetic ordering in good agreement with that observed in the AC susceptibility measurements. Fig. 2 shows the thermal evolution of the lattice parameters and cell volume as obtained from the neutron diffractograms. The a and c unit-cell parameters present a different thermal behavior. Moreover, the cell volume smoothly decreases with decreasing temperature, and remains constant below 50 K. No phase segregation was found even with the temperature down to 3 K. The result was similar to that of a previous report [17].

The development of magnetic correlations with temperature is shown in Fig. 3. The spin arrangements for the in-plane Mn ions are also plotted in the inset. The data

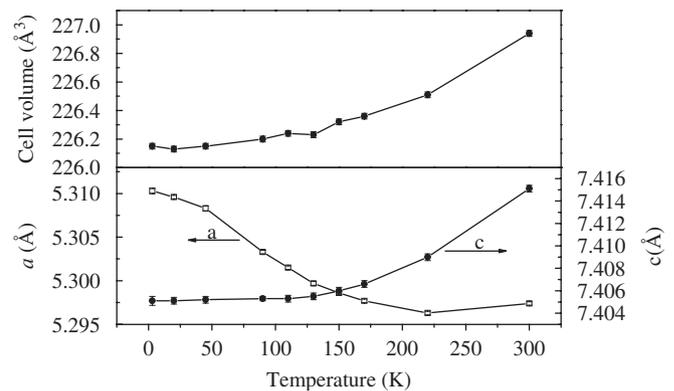


Fig. 2. Thermal evolution of the a and c lattice parameters and cell volume of $\text{Tb}_{0.85}\text{Na}_{0.15}\text{MnO}_{3-y}$.

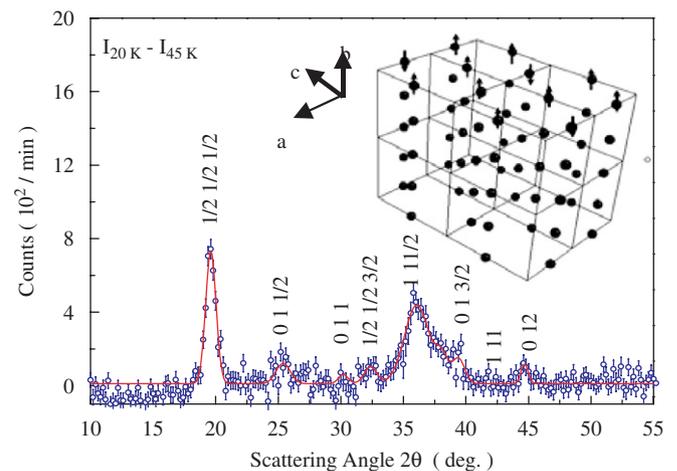


Fig. 3. Magnetic contribution as obtained by subtracting the high temperature (45 K) from the low temperature (20 K) diffraction data. The spin arrangements of the in-plane Mn ions are also plotted in the inset.

were collected using a BT-9 triple-axis spectrometer at NIST, and the solid curves show the fits of the data to short-range scattering profiles convoluted with the Gaussian instrumental resolution. The diffraction peaks may be considered using the $[\frac{1}{2}\frac{1}{2}\frac{1}{2}]$ wave vector, with widths that are much broader than the instrumental resolution, revealing an antiferromagnetic short-range order with a correlation length of $\xi \approx 60 \text{ \AA}$ for the Tb spins. By comparison with $T_{\text{Tb}} \approx 7 \text{ K}$ and $\xi \approx 140 \text{ \AA}$ for the undoped TbMnO_3 compound [10], the reduction in ξ value can be explained by assuming the decrease in Tb content which may create more oxygen vacancies, and then weaken the correlations among the Tb ions.

4. Conclusions

In this study, we can confirm that the oxygen vacancies in $\text{Tb}_{0.85}\text{Na}_{0.15}\text{MnO}_{3-y}$ [$y \approx 0.15(2)$] play an important role in controlling the crystal structure and magnetic properties.

Acknowledgments

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